

## 14.7 Carboxylic acids

01. 0620\_m21\_qp\_22 Q: 38

The formula of an ester is  $\text{CH}_3\text{CH}_2\text{CH}_2\text{COOCH}_2\text{CH}_2\text{CH}_3$ .

Which acid and alcohol react together to make the ester?

	acid	alcohol
<b>A</b>	butanoic acid	butanol
<b>B</b>	butanoic acid	propanol
<b>C</b>	propanoic acid	butanol
<b>D</b>	propanoic acid	propanol

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02. 0620\_s21\_qp\_21 Q: 34

Compound Z contains carbon, hydrogen and oxygen.

Molecules of compound Z have four hydrogen atoms and two carbon atoms.

Compound Z can be made by oxidation of an alcohol.

What is compound Z?

- A** ethene
- B** ethanol
- C** ethanoic acid
- D** methyl methanoate

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03. 0620\_s21\_qp\_22 Q: 34

Which statement about ethanol is **not** correct?

- A** Ethanol can be made by fermentation.
  - B** Ethanol is oxidised to make ethanoic acid.
  - C** Ethanol reacts with oxygen exothermically, making it a good fuel.
  - D** Ethanol reacts with propanoic acid to make propyl ethanoate.
-

14.7. CARBOXYLIC ACIDS

04.0620\_s21\_qp\_22 Q: 37

37 Which statements about ethanoic acid are correct?

- 1 It is a strong acid.
- 2 It reacts with ethanol to form an ester.
- 3 It has the formula  $\text{CH}_3\text{COOH}$ .

**A** 1 and 2 only    **B** 1 and 3 only    **C** 2 and 3 only    **D** 1, 2 and 3

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05.0620\_s21\_qp\_23 Q: 34

Ethanol is reacted with acidified potassium manganate(VII).

Which row describes the type of reaction and the type of organic compound formed?

	type of reaction	organic compound
<b>A</b>	oxidation	carboxylic acid
<b>B</b>	oxidation	alkene
<b>C</b>	dehydration	carboxylic acid
<b>D</b>	dehydration	alkene

06. 0620\_w21\_qp\_21 Q: 39

Which statement about aqueous ethanoic acid is correct?

- A It reacts with magnesium to produce a salt and hydrogen.
- B It reacts with sodium hydroxide to produce a salt and hydrogen.
- C It reacts with ammonium salts to produce ammonia.
- D It turns red litmus blue.

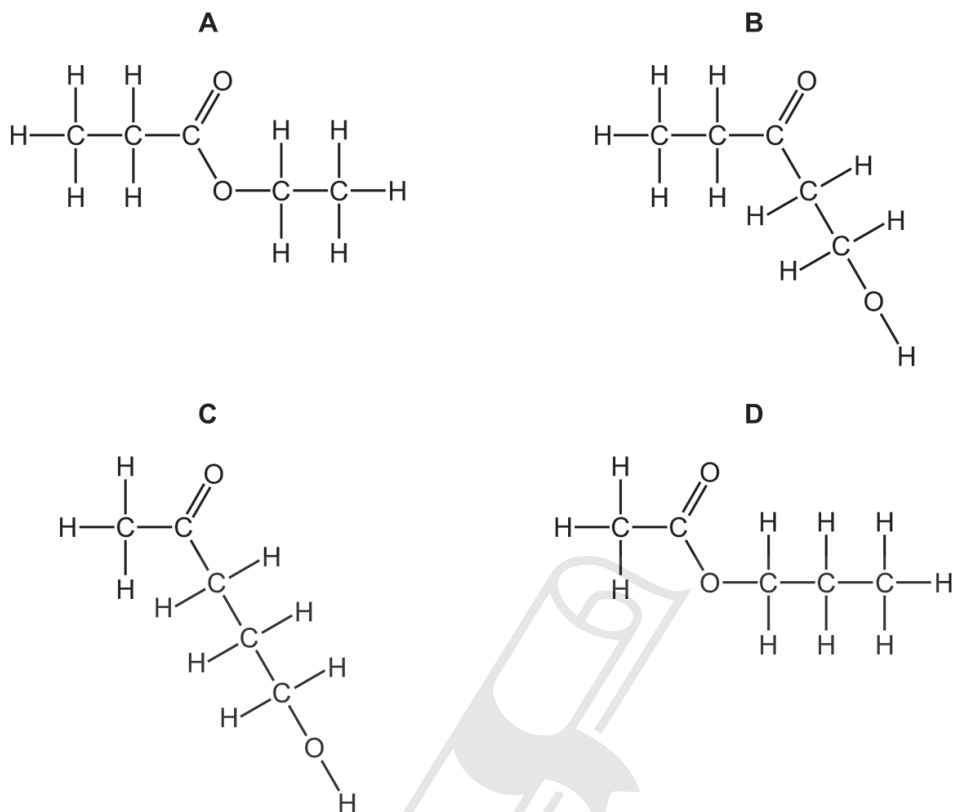


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14.7. CARBOXYLIC ACIDS

07. 0620\_w21\_qp\_22 Q: 33

What is the structure of the ester formed from ethanoic acid and propanol?



08. 0620\_m20\_qp\_22 Q: 38

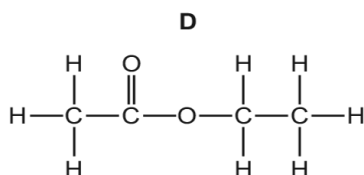
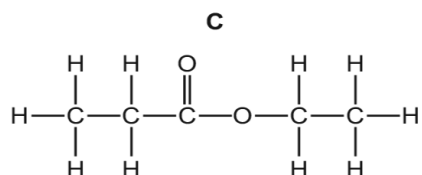
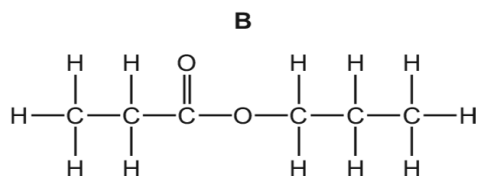
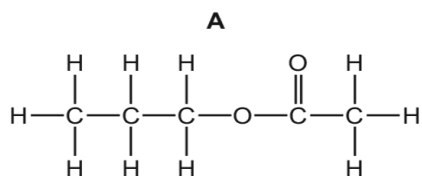
Ethanoic acid is a typical carboxylic acid.

Which statement about ethanoic acid is correct?

- A** It can be oxidised to produce ethanol.
- B** It is a proton acceptor.
- C** It is fully dissociated in water.
- D** It reacts with ethanol to produce ethyl ethanoate and water.

09. 0620\_m20\_qp\_22 Q: 39

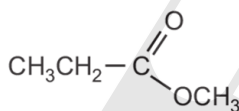
Which structure represents the ester made from ethanoic acid and propanol?



10. 0620\_p20\_qp\_20 Q: 39

Esters are made by reacting an alcohol with a carboxylic acid.

Which acid and alcohol react together to form the following ester?



- A** propanoic acid and ethanol  
**B** propanoic acid and methanol  
**C** ethanoic acid and ethanol  
**D** ethanoic acid and methanol

11. 0620\_s20\_qp\_21 Q: 38

Some properties of an organic compound J are listed.

- It is a liquid at room temperature.
- It is soluble in water.
- A solution of J reacts with calcium carbonate to form carbon dioxide.
- A solution of J has a pH of 3.

In which homologous series does J belong?

- A** alkane  
**B** alkene  
**C** alcohol  
**D** carboxylic acid

14.7. CARBOXYLIC ACIDS

12. 0620\_s20\_qp\_22 Q: 38

An organic compound, P, reacts with zinc to produce a gas, Q.

What are P and Q?

	P	Q
<b>A</b>	ethanoic acid	carbon dioxide
<b>B</b>	ethanoic acid	hydrogen
<b>C</b>	ethanol	carbon dioxide
<b>D</b>	ethanol	hydrogen

13. 0620\_s20\_qp\_23 Q: 38

A small quantity of a solid chemical is added to a large excess of aqueous ethanoic acid.

No bubbles of gas are seen and the solid dissolves to give a colourless solution.

What was the solid chemical?

- A** calcium hydroxide
- B** copper(II) oxide
- C** magnesium
- D** sodium carbonate

14. 0620\_w20\_qp\_21 Q: 39

Which statements about aqueous ethanoic acid are correct?

- 1 It is an alkane.
- 2 It reacts with sodium carbonate to form carbon dioxide.
- 3 It changes the colour of litmus solution from blue to red.
- 4 It is a hydrocarbon.

- A** 1 and 2      **B** 1 and 4      **C** 2 and 3      **D** 3 and 4

15. 0620\_w20\_qp\_23 Q: 37

Propanol is oxidised by acidified potassium manganate(VII) in a similar way to ethanol.

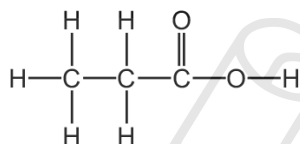
Which compound is produced by the oxidation of propanol with acidified potassium manganate(VII)?

- A**  $\text{CH}_3\text{CH}_2\text{OH}$   
**B**  $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$   
**C**  $\text{CH}_3\text{COOH}$   
**D**  $\text{CH}_3\text{CH}_2\text{COOH}$

16. 0620\_m19\_qp\_22 Q: 39

The structure of a compound, G, is shown.

G is in the same homologous series as ethanoic acid.



Which row describes some of the properties of an aqueous solution of G?

	produces a gas with magnesium	turns methyl orange yellow
<b>A</b>	no	yes
<b>B</b>	no	no
<b>C</b>	yes	no
<b>D</b>	yes	yes

17. 0620\_s19\_qp\_21 Q: 38

Which statements about aqueous ethanoic acid are correct?

- Ethanoic acid contains the functional group  $-\text{COOH}$ .
- Ethanoic acid reacts with carbonates to produce hydrogen.
- Ethanoic acid turns Universal Indicator paper blue.
- Ethanoic acid has a pH lower than pH 7.

- A** 1 and 2      **B** 1 and 3      **C** 1 and 4      **D** 2 and 4

14.7. CARBOXYLIC ACIDS

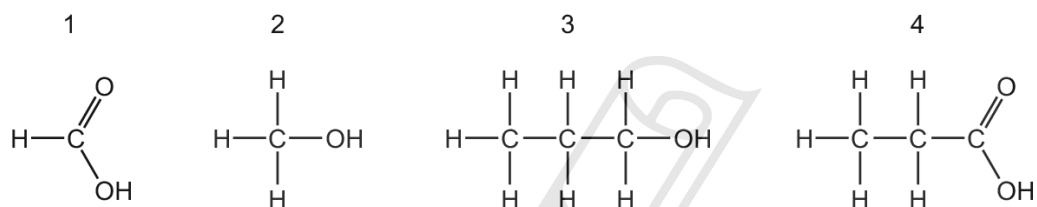
18. 0620\_s19\_qp\_22 Q: 38

What are the properties of aqueous ethanoic acid?

	decolourises bromine water	reacts with calcium carbonate to make carbon dioxide	turns damp red litmus blue
<b>A</b>	✓	✓	✗
<b>B</b>	✓	✗	✓
<b>C</b>	✗	✓	✗
<b>D</b>	✗	✗	✓

19. 0620\_s19\_qp\_22 Q: 39

The structures of four molecules are shown.



Which molecules react together to form the ester propyl methanoate?

- A** 1 and 2            **B** 1 and 3            **C** 2 and 4            **D** 3 and 4

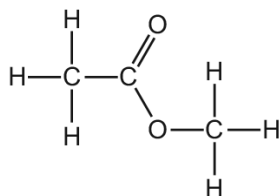
20. 0620\_s19\_qp\_23 Q: 38

Which statement about aqueous ethanoic acid is correct?

- A** It reacts with metal carbonates to form salts, hydrogen and water.  
**B** It reacts with metal oxides to form salts and oxygen.  
**C** It reacts with reactive metals to form salts and hydrogen.  
**D** It turns damp red litmus paper blue.

21. 0620\_s19\_qp\_23 Q: 39

The structure of ester W is shown.



Which row gives the names of ester W and the carboxylic acid and alcohol from which it is made?

	name of ester W	carboxylic acid	alcohol
<b>A</b>	ethyl methanoate	ethanoic acid	methanol
<b>B</b>	ethyl methanoate	methanoic acid	ethanol
<b>C</b>	methyl ethanoate	ethanoic acid	methanol
<b>D</b>	methyl ethanoate	methanoic acid	ethanol

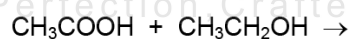
22. 0620\_m18\_qp\_22 Q: 39

Which substances react together to form ethyl propanoate?

- A** ethanoic acid and propanol
- B** ethanol and propene
- C** ethene and propanol
- D** propanoic acid and ethanol

23. 0620\_s18\_qp\_22 Q: 39

What is the name of the organic product of the reaction shown?



- A** ethyl ethanoate
- B** ethyl methanoate
- C** methyl ethanoate
- D** methyl propanoate

14.7. CARBOXYLIC ACIDS

24. 0620\_s18\_qp\_23 Q: 39

Which esters have the molecular formula  $C_5H_{10}O_2$ ?

- 1 ethyl propanoate
  - 2 propyl ethanoate
  - 3 butyl methanoate
  - 4 methyl butanoate
- A** 1, 2, 3 and 4  
**B** 1, 2 and 3 only  
**C** 1 and 2 only  
**D** 3 and 4 only
- 

25. 0620\_w18\_qp\_21 Q: 36

Which two compounds are molecules which both contain a double bond?

- A** ethane and ethanoic acid
  - B** ethane and ethanol
  - C** ethene and ethanoic acid
  - D** ethene and ethanol
- 

26. 0620\_w18\_qp\_21 Q: 38

When the alcohol  $CH_3CH_2CH_2OH$  reacts with the carboxylic acid  $CH_3CH_2CH_2COOH$  an ester is formed.

What is the name and structural formula of this ester?

	name	structural formula
<b>A</b>	butyl propanoate	$CH_3CH_2COOCH_2CH_2CH_2CH_3$
<b>B</b>	butyl propanoate	$CH_3CH_2CH_2COOCH_2CH_2CH_3$
<b>C</b>	propyl butanoate	$CH_3CH_2COOCH_2CH_2CH_2CH_3$
<b>D</b>	propyl butanoate	$CH_3CH_2CH_2COOCH_2CH_2CH_3$

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27. 0620\_w18\_qp\_21 Q: 39

A solution of ethanol and water is left to stand in an open beaker in a warm room for three weeks.

Which statement explains what happens to the ethanol in the solution?

- A The ethanol is dehydrated to ethene.
  - B The ethanol is hydrolysed to ethene.
  - C The ethanol is oxidised to ethanoic acid.
  - D The ethanol is reduced to ethanoic acid.
- 

28. 0620\_w18\_qp\_22 Q: 39

Which reaction can be used to make ethanoic acid?

- A oxidation of ethanol
  - B oxidation of ethene
  - C reduction of ethanol
  - D reduction of ethene
- 

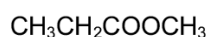
29. 0620\_w18\_qp\_23 Q: 39

Which statement about ethanoic acid is correct?

- A It contains a  $-C_2H_5$  group.
  - B It is a strong acid.
  - C It is formed by the reduction of ethanol.
  - D It reacts with alcohols to form esters.
- 

30. 0620\_m17\_qp\_22 Q: 38

The structural formula of an organic compound is shown.



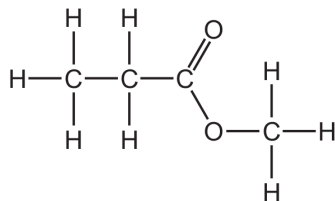
What is the name of this compound?

- A butanoic acid
  - B ethyl ethanoate
  - C methyl propanoate
  - D propyl methanoate
-

14.7. CARBOXYLIC ACIDS

31. 0620\_s17\_qp\_21 Q: 39

The structure of an ester is shown.



Which row is correct?

	name of ester	names of the carboxylic acid and the alcohol used to form the ester
<b>A</b>	methyl propanoate	methanoic acid and propanol
<b>B</b>	methyl propanoate	methanol and propanoic acid
<b>C</b>	propyl methanoate	methanoic acid and propanol
<b>D</b>	propyl methanoate	methanol and propanoic acid

32. 0620\_s17\_qp\_22 Q: 39

The formula of an ester is  $\text{CH}_3\text{CH}_2\text{CH}_2\text{COOCH}_2\text{CH}_2\text{CH}_3$ .

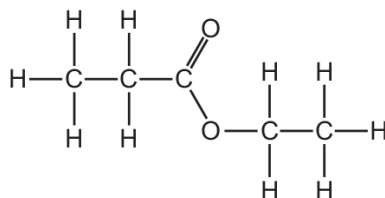
Which acid and alcohol react together to make the ester?

	acid	alcohol
<b>A</b>	butanoic acid	butanol
<b>B</b>	butanoic acid	propanol
<b>C</b>	propanoic acid	butanol
<b>D</b>	propanoic acid	propanol

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33. 0620\_s17\_qp\_23 Q: 39

The structure of an ester is shown.



Which alcohol and carboxylic acid produce this ester?

	alcohol	carboxylic acid
<b>A</b>	ethanol	ethanoic acid
<b>B</b>	ethanol	propanoic acid
<b>C</b>	propanol	ethanoic acid
<b>D</b>	propanol	propanoic acid

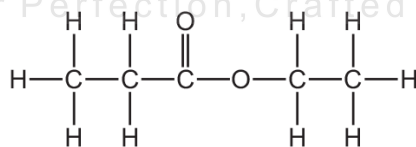
34. 0620\_w17\_qp\_21 Q: 39

Which pair of compounds can be used to prepare  $\text{CH}_3\text{CH}_2\text{COOCH}_2\text{CH}_3$ ?

- A** ethanoic acid and ethanol
- B** ethanoic acid and propanol
- C** propanoic acid and ethanol
- D** propanoic acid and propanol

35. 0620\_w17\_qp\_22 Q: 39

The structure of an ester is shown.



Which substances react to form this ester?

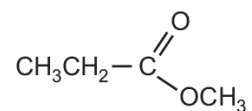
- A** ethanol and ethanoic acid
- B** ethanol and propanoic acid
- C** propanol and ethanoic acid
- D** propanol and propanoic acid



38. 0620\_p16\_qp\_20 Q: 39

Esters are made by reacting an alcohol with a carboxylic acid.

Which acid and alcohol react together to form the following ester?



- A** propanoic acid and ethanol  
**B** propanoic acid and methanol  
**C** ethanoic acid and ethanol  
**D** ethanoic acid and methanol

39. 0620\_s16\_qp\_21 Q: 37

Which compounds contain the same number of carbon, hydrogen and oxygen atoms?

W	X	Y	Z
ethyl methanoate	methyl ethanoate	methyl methanoate	ethyl ethanoate

- A** W and X      **B** W and Y      **C** X and Z      **D** Y and Z

40. 0620\_s16\_qp\_22 Q: 37

An ester is formed when a carboxylic acid reacts with an alcohol.

Which ester is formed when propanoic acid and ethanol react?

- A**  $\text{CH}_3\text{CO}_2\text{CH}_2\text{CH}_3$   
**B**  $\text{CH}_3\text{CO}_2\text{CH}_2\text{CH}_2\text{CH}_3$   
**C**  $\text{CH}_3\text{CH}_2\text{CO}_2\text{CH}_3$   
**D**  $\text{CH}_3\text{CH}_2\text{CO}_2\text{CH}_2\text{CH}_3$

SN	Paper	Q. No.	Answer
01	0620_m21_qp_22	38	B
02	0620_s21_qp_21	34	C
03	0620_s21_qp_22	34	D
04	0620_s21_qp_22	37	C
05	0620_s21_qp_23	34	A
06	0620_w21_qp_21	39	A
07	0620_w21_qp_22	33	D
08	0620_m20_qp_22	38	D
09	0620_m20_qp_22	39	A
10	0620_p20_qp_20	39	B
11	0620_s20_qp_21	38	D
12	0620_s20_qp_22	38	B
13	0620_s20_qp_23	38	A
14	0620_w20_qp_21	39	C
15	0620_w20_qp_23	37	D
16	0620_m19_qp_22	39	C
17	0620_s19_qp_21	38	C
18	0620_s19_qp_22	38	B
19	0620_s19_qp_22	39	C
20	0620_s19_qp_23	38	C
21	0620_s19_qp_23	39	C
22	0620_m18_qp_22	39	D
23	0620_s18_qp_22	39	A
24	0620_s18_qp_23	39	A
25	0620_w18_qp_21	36	C
26	0620_w18_qp_21	38	D
27	0620_w18_qp_21	39	C
28	0620_w18_qp_22	39	A
29	0620_w18_qp_23	39	D
30	0620_m17_qp_22	38	C
31	0620_s17_qp_21	39	B
32	0620_s17_qp_22	39	B
33	0620_s17_qp_23	39	B
34	0620_w17_qp_21	39	C
35	0620_w17_qp_22	39	B
36	0620_w17_qp_23	39	D
37	0620_m16_qp_22	38	D
38	0620_p16_qp_20	39	B
39	0620_s16_qp_21	37	A
40	0620_s16_qp_22	37	D